

Journal of Pharmaceutical Research International

33(52B): 276-292, 2021; Article no.JPRI.78061 ISSN: 2456-9119 (Past name: British Journal of Pharmaceutical Research, Past ISSN: 2231-2919, NLM ID: 101631759)

Synthesis, Characterization and Evaluation of Antidiabetic Activity of Novel Pyrazoline Fused Indole Derivatives

Nagesh Vaddiraju ^{a*}, M. Ajitha ^{b†} and K. Rajnarayana ^{c#}

^a Department of Pharmaceutical Chemistry, G. Pulla Reddy College of Pharmacy, Hyd-28, T.S, India. ^b CPS, IST, JNTU, Hyderabad, T.S, India. ^c Biotechnica Pvt Limited, Kukatpally, Hyderabad, T.S, India.

Authors' contributions

This work was carried out in collaboration among all authors. All authors read and approved the final manuscript.

Article Information

DOI: 10.9734/JPRI/2021/v33i52B33627 <u>Editor(s)</u>: (1) Rafik Karaman, Al-Quds University, Palestine. <u>Reviewers:</u> (1) R. Ganesamoorthy, Tel Aviv University, Israel. (2) Asif Mahmood, University of Sargodha, Pakistan. (3) S. Ahalya, Seethalakshmi Ramaswami College, India. Complete Peer review History, details of the editor(s), Reviewers and additional Reviewers are available here: <u>https://www.sdiarticle5.com/review-history/78061</u>

Original Research Article

Received 25 September 2021 Accepted 01 December 2021 Published 02 December 2021

ABSTRACT

The primary purpose of this research work is to synthesize, characterize and biological evaluation of novel pyrazoline fused indole derivatives lead to creating a new molecular frame work. **Methodology:** In the present study, the new series of novel pyrazoline fused indole derivatives were synthesized from from indole and substituted acetophenone by the 4 step process. In the first step indole and dimethyl formamide were coupled by using phosphorous oxychloride and NaOH to prepare the compound 1 Indole-3-aldehyde. In the second step compound 1 was condensed with substituted aetophenone to synthesis the compound 2 chalcones (a-h). In the third step chalcones 2(a-h) were coupled with semicarbazide or thiosemicarbazide to synthesis the compound 3(a-p). In the final step compound 3(a-p) were coupled with indole-3-aldehyde to prepare the final product of R-substitutedN-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carboxamide and R-substitutedN-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide 4(a-p).

[†]Professor;

[#]General Manager;

^{*}Corresponding author: E-mail: nagesh.vaddiraju@gmail.com;

Results: The chemical structures of the synthesized compounds were characterized by means of IR, Mass and NMR spectroscopy. The compounds were screened for anti-diabetic activity by Invitro and In-vivo methods. In In-vivo method 4a, 4m have exhibited moderate anti-diabetic activity as that of standard drug, glibenclamide. In In-vitro method 4a, 4e & 4m have shows moderate anti-diabetic activity as that of reference standard, acarbose.

Conclusion: The synthesized novel pyrazoline fused indole derivatives have moderate antidiabetic activity as that of standard drug by In-vitro and In-vivo methods. These compounds can be further exploited to get the potent lead compound.

Keywords: Pyrazoline; indole; indole-3-aldehyde and antidiabetic activity

1. INTRODUCTION

1.1 Pyrazolines

Pyrazolines are the reduced form of the pyrazole moiety having one endocyclic double bond. There are 3 isomers of pyrazoline moiety based on the position of the endocyclic double bond. They are the one of the most studied group among the azole family. Pyrazolines have been proved to possess analgesic, anti-inflammatory, antipyretic, antidiabetic, antibacterial, antifungal, parasitic. antitubercular, anti insecticidal. cytotoxic activities [1]. Many methods have been described in the literature for the synthesis of substituted pyrazolines. Some of the methods synthesis from chalcones, from aryl are hydrazines by reacting with 3-butynol, from alkyl dihalides by reacting with primary amine [2], from cyclocondensation of α , β -unsaturated carbonyl compounds with hydrazine or arvlhydrazines [3] and subsequent dehydrogenation. In order to develop an efficient synthetic approach to the various 3 and 5 substituted-1H-pyrazolines. Although these methods were found many applications in synthesizing novel pyrazolines, there is still a great deal of work remaining to enable the development of efficient protocols for structurally different compounds and to make more practical by these reactions usina inexpensive and easily available starting materials.

1.2 Indole

Indole is also known as benzopyrrole which contains benzenoid nucleus and has $10 \text{ }\pi$ -electrons (two from lone pair on nitrogen and double bonds provide eight electrons) which makes them aromatic in nature. Similar to the benzene ring, electrophilic substitution occurs readily on indole due to excessive π -electrons delocalization [4]. Indole is an important heterocyclic system that provides the skeleton to lysergic acid diethylamide (LSD), strychnine and

alkaloid obtained from plants. Physically, they are crystalline colorless in nature with specific odors. The addition of the indole nucleus to medicinal compounds that is biologically active it pharmacophore made an important heterocyclic compound having broad-spectrum biological activities [5]. Due to this, researchers took interest to synthesize various scaffolds of indole for screening different pharmacological activities. Various natural compounds contain indole as parent nucleus for example tryptophan. Indole-3-acetic acid is a plant hormone produced by the degradation of tryptophan in higher plants. Derivatives of indole are of wide interest because diverse biological of their and clinical applications. Here, we have tried to summarize the important pharmacological activity of indole derivatives [6]. Indole derivatives possess various biological activities, i.e., antiviral [7], antiinflammatory [8], anticancer [9], antiHIV [10], antioxidant [11], antimicrobial [12], antitubercular antidiabetic [13], [14], antimalarial [15], anticholinesterase activities [16], etc.

Many methods have been described in the literature for the synthesis of substituted indoles. Some of the methods are synthesis from Fischer indole synthesis [17], aryl bromides and allyl alcohols [18], N-nitrosoanilines with alkynes [19], o-bromonitrobenzenes with various vinvl Grignard reagents [20] and o-nitrobenzyl cyanides with boronic acids [21]. Although these methods were found many applications in synthesizing novel indoles, there is still a great deal of work remaining to enable the development of efficient protocols for structurally different compounds and to make these reactions more practical by using inexpensive and easily available starting materials.

2. MATERIALS AND METHODS

All the required chemicals used were obtained from Aldrich and Sd-fine chemicals. All the solvents used were of laboratory grade. Each reaction was monitored by TLC by using appropriate solvent system, which was selected by trial and error method. Precoated TLC plates (0.25mm silica gel) were obtained from E. Merck. All the synthesized compounds were purified by recrystallization. Melting points were determined on Fisher Johns melting point apparatus and they were uncorrected. All the H¹- NMR spectra were recorded on sophisticated multinuclear FT-NMR spectrometer model Avance-II (Bruker) using CDCl₃ and DMSO-d₆ as solvents, tetra methyl silane (TMS) as internal standard.

Mass spectra of the compounds were recorded on Mass spectrometer model Agilent 1100 series, and the method used is ESI method. And they were reported in m/z value as molecular ion peak. IR spectra were recorded on Nexus 670 FTIR thermonicolet instrument by KBr disc method.

2.1 The Final Compounds were Synthesized as Given Below

2.1.1 Scheme

Step 1:- Synthesis of Indole-3-aldehyde Step 2:- Synthesis of R-substituted 3-(1Hindol-3-yl)-1-phenylprop-2-en-1-one Step 3:- Synthesis of R-substituted 5-(1H-

indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carboxamide and R-substituted 5-(1Hindol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide

Step 4:- Synthesis of R-substituted N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carboxamide and R-substituted N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide

2.1.1.1 Step 1

Synthesis of Indole-3-aldehyde: In threenecked RBF add 288ml (274 g., 3.74 moles) of freshly distilled dimethylformamide and contents are cooled in an ice-salt bath for about 30 min, and 86 ml. (144 g., 0.94 mole) of freshly distilled POCl₃ is subsequently added with stirring to the dimethylformamide over a period of 30 min. Then added a solution of 100 g. (0.85 mole) of indole in 100 ml. (95 g., 1.3 moles) of dimethylformamide to the yellow solution over a period of 1 hour during which time the temperature should not rise above 10°C. The syrup is stirred efficiently at this temperature for 1 hour. At the end of the reaction period, 300 g. of

crushed ice is added. This solution is transferred with 100 ml. of water to a three-necked flask containing 200 g. of crushed ice and fitted efficient mechanical stirrer with an and separatory funnel containing a solution of 375 g. (9.4 moles) of NaOH in 1I of water. The aqueous base is added dropwise with stirring until about one-third of it has been added. The remaining two-thirds is added rapidly with efficient stirring and the resulting suspension is heated rapidly to the boiling point and allowed to cool to room temperature, after which it is placed in a refrigerator overnight. The precipitate is collected on a filter and resuspended in 1 l. of water. Most of the inorganic material dissolves, and the product is then collected on a filter, washed with three 300-ml. portions of water and air-dried, to get indole-3-aldehyde, recrystallized from ethanol if desired.

2.1.1.2 Step 2

Synthesis of R-substituted 3-(1H-indol-3-vl)-1phenylprop-2-en-1-one: In a 500ml of bolt head flask take a solution of 22gm of sodium hydroxide in 200ml of water and 100gm (122.5ml) rectified spirit. Immerse the flask in a bath of a crushed ice, then add (0.43mol) of freshly distilled acetophenone, start the stir and (0.43mol) Indole-3then add of pure carboxaldehyde. Keep the temperature of the mixture at about 25°C (limits are 15 -30°C) and stir vigorously until the mixture is so thick that stirring is no longer effective (2-3hours). Remove the stirrer and leave the reaction mixture in an ice-chest or refrigerator overnight. Filter the product with suction on a Buchner funnel or a sintered glass funnel and wash with cold water until the washings are neutral to litmus and then with 20ml of ice-cold rectified spirit. Recrystallise from rectified spirit warmed to 50°C (about 5ml per gram).

2.1.1.3 Step 3

Synthesis of R-substituted 5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-

carboxamide and R-substituted 5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-

carbothioamide: To the solution of chalcone derivatives (0.01mol) and semicarbazide or thiosemicarbazide (0.012mol) in 25 ml of ethanol, a solution of sodium hydroxide (0.025mol) in 5ml of water was added and refluxed for 8 hour. The products were poured into crushed ice and the solid mass which separated out was filtered dried and recrystallized from appropriate solvents.

2.1.1.4 Step 4

Synthesis of R-substituted N-((1H-indol-3yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5dihydro-1H-pyrazole-1-carboxamide and Rsubstituted N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1Hpyrazole-1-carbothioamide: To the solution of pyrazole derivatives (0.01mol) and Indole-3carboxaldehyde (0.01mol) in 25 ml of ethanol and add 2-5ml of glacial acetic acid and refluxed for 2 hour. The products were poured into crushed ice and the solid mass which separated out was filtered dried and recrystallized from

2.2 Anti-Diabetic Activity

appropriate solvents.

2.2.1 In vivo anti-diabetic activity

Experimental Animals: Male Wistar rats (170–220 g) were used to study the antidiabetic activity. Animals were housed in standard laboratory conditions (temperature $22 \pm 2^{\circ}C$ and humidity $45 \pm 5^{\circ}C$ with 12h day: 12h night cycle).

Acute Toxicity Studies: The acute toxicity study of synthesized compounds was performed as per guidelines of the Committee for the Purpose of Control and Supervision of Experiments on Animals (CPCSEA). The synthesized compounds were administrated to rats by gavage using a stomach tube in increasing dose levels of 100, 500 and 1000 mg/kg b.wt, respectively [22]. All animals were observed for gross behavioral, neurological, autonomic and toxic effects at short intervals of time for 5 h after administration and then for next 24 h. Food consumption and body weights were recorded daily for 21 days.

Induction of Diabetes: Diabetes was induced by a single intraperitoneal injection of STZ (55 mg/kg b.wt), freshly prepared in 0.1M sodium citrate buffer (pH 4.5) after overnight fasting [23]. Animals were fed with 5% glucose solution for 12 h to avoid hypoglycaemia. On the 4th day of STZ administration, blood glucose level was measured through Glucometer and the rats with moderate diabetes, having hyperglycemia (blood glucose range of above 250 mg/dl) were considered as diabetic and were employed in the study.

Experimental Design: A total of 30 rats (6 normal; 24 STZ-diabetic rats) were assigned to the study. The rats were randomly divided into five groups of six animals each. Group 1: Normal

control rats (NC) received vehicle only (1% CMC; 2 ml/kg b.wt), Group 2: Diabetic control rats (DC), Group 3 and 4: Diabetic rats received synthesized compounds 4a and 4g at the dose of 50 mg/kg b.wt, respectively, Group 5: Positive control received a reference standard drug Glibenclamide (5 mg/kg b.wt). All treatments were given orally after the 4th day of STZ administration (except normal control) for 21 days. The body weight was recorded initially and after the end of the treatment. Blood was withdrawn from the tail vein each time blood glucose level was measured by Glucometer (one touch, Johnson & Johnson) on 0, 7, 14 and 21 day of the study.

Estimation of blood glucose and glycosylated hemoglobin (HbA1c): After the completion of treatment, the rats were fasted overnight and blood samples for fasting blood glucose and glycosylated hemoglobin (HbA1c) were obtained from the tail vein under mild ether anesthesia. Fasting blood glucose was measured by Glucometer (one touch, Johnson & Johnson) and glycosylated hemoglobin (HbA1c) was estimated using the method of Nayak and Pattabiraman [24].

2.2.2 In vitro anti-diabetic activity

In vitro α-amylase Inhibition Assay: The αamylase inhibitory activity was determined by using soluble starch as a substrate in a colorimetric reaction by the method of Bernfield [23]. α-amylase was dissolved in phosphate buffer saline (PBS, 0.02 mol/L, pH 6.8) at a mg/mL. concentration of 0.1 Various concentrations of sample solutions (0.25 mL) were mixed with α -amylase solution (0.25 mL) and incubated at 37°C for 5 min. Then the reaction was initiated by adding 0.5 mL 1.0% (w/v) starch substrate solution to the incubation medium. After incubation at 37°C for 3 min. the reaction was stopped by adding 0.5 mL DNS reagent (1% Dinitrosalicylic acid, 0.05% Na₂SO₃ and 1% NaOH solution) to the reaction mixture and boiling at 100°C for 5 min. After cooling to room temperature, the absorbance (Abs) at 540 nm was recorded by a spectrophotometer [25].

In vitro α - glucosidase Inhibition Assay: The inhibitory activity was determined by incubating a solution of starch substrate (2 % w/v maltose or sucrose) 1 ml with 0.2 M Tris buffer pH 8.0 and various concentration of control (Acarbose std. drug) and the synthesized compound(s) for 5 min at 37°C. The reaction was initiated by adding 1

ml of alpha-glucosidase enzyme $(1\mu g/ml)$ to it followed by incubation for 40 min at 35°C. Then the reaction was terminated by the addition of 2 ml of 6N HCl. Then the intensity of the colour was measured at 540 nm [25].

The percentage inhibition of α -amylase and α -glucosidase was calculated using the following formula:

 $\frac{\text{Percentage inhibition} =}{\frac{\text{Absorbance of control} - \text{Absorbance of test}}{\text{Absorbance of control}} \times 100$

3. RESULTS AND DISCUSSION

The synthesis and biological evaluation of novel pyrazoline fused indole derivatives lead to creating a new molecular frame work. Sixteen

3.1 Scheme

novel pyrazoline fused indole derivatives from indole and substituted acetophenone were prepared by following steps. In the first step indole and dimethyl formamide were coupled by using phosphorous oxychloride and NaOH to prepare the compound 1. In the second step compound 1 was condensed with substituted aetophenone to synthesis the compound 2(a-h). In the third step chalcones 2(a-h) were coupled with semicarbazide or thiosemicarbazide to synthesis the compound 3(a-p). In the final step compound 3(a-p) were coupled with indole-3aldehyde to prepare the final product of RsubstitutedN-((1H-indol-3-yl)methylene)-5-(1Hindol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1carboxamide and R-substitutedN-((1H-indol-3yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5dihydro-1H-pyrazole-1-carbothioamide 4(a-p).



R= -H, -CH3, -C2H5, -OCH3, -NO2, -Cl, -OH, -N(CH3)2 X= -O, -S

Comp	-X	-R	IUPAC Name
4a	-0	-H	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-
		.	1H-pyrazole-1-carboxamide
4b	-0	$-CH_3$	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(p-tolyl)-4,5-dihydro-
			1H-pyrazole-1-carboxamide
4c	-0	$-C_2H_5$	N-((1H-indol-3-yl)methylene)-3-(4-ethylphenyl)-5-(1H-indol-3-yl)-4,5-
	-		dihydro-1H-pyrazole-1-carboxamide
4d	-0	-OCH ₃	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(4-methoxyphenyl)-
			4,5-dihydro-1H-pyrazole-1-carboxamide
4e	-0	-NO ₂	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(4-nitrophenyl)-4,5-
			dihydro-1H-pyrazole-1-carboxamide
4f	-0	-Cl	N-((1H-indol-3-yl)methylene)-3-(4-chlorophenyl)-5-(1H-indol-3-yl)-
			4,5-dihydro-1H-pyrazole-1-carboxamide
4g	-0	-OH	N-((1H-indol-3-yl)methylene)-3-(4-hydroxyphenyl)-5-(1H-indol-3-yl)-
			4,5-dihydro-1H-pyrazole-1-carboxamide
4h	-0	-N(CH ₃) ₂	N-((1H-indol-3-yl)methylene)-3-(4-(dimethylamino)phenyl)-5-(1H-
			indol-3-yl)-4,5-dihydro-1H-pyrazole-1-carboxamide
4i	-S	-H	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-
			1H-pyrazole-1-carbothioamide
4j	-S	-CH ₃	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(p-tolyl)-4,5-dihydro-
			1H-pyrazole-1-carbothioamide
4k	-S	$-C_2H_5$	N-((1H-indol-3-yl)methylene)-3-(4-ethylphenyl)-5-(1H-indol-3-yl)-4,5-
			dihydro-1H-pyrazole-1-carbothioamide
41	-S	-OCH ₃	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(4-methoxyphenyl)-
			4,5-dihydro-1H-pyrazole-1-carbothioamide
4m	-S	-NO ₂	N-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-(4-nitrophenyl)-4,5-
			dihydro-1H-pyrazole-1-carbothioamide
4n	-S	-CI	N-((1H-indol-3-yl)methylene)-3-(4-chlorophenyl)-5-(1H-indol-3-yl)-
			4,5-dihydro-1H-pyrazole-1-carbothioamide
40	-S	-OH	N-((1H-indol-3-yl)methylene)-3-(4-hydroxyphenyl)-5-(1H-indol-3-yl)-
			4,5-dihydro-1H-pyrazole-1-carbothioamide
4р	-S	-N(CH ₃) ₂	N-((1H-indol-3-yl)methylene)-3-(4-(dimethylamino)phenyl)-5-(1H-
			indol-3-yl)-4,5-dihydro-1H-pyrazole-1-carbothioamide

Table-1. List of synthesized compounds 4 (a-p)

Table 2. Physical data of R-substitutedN-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-
4,5-dihydro-1H-pyrazole-1-carboxamideandN-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-
phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide

Comp	-X	-R	Mol formula	Mol.wt	M.P (⁰C)	Yield (%)
4a	-0	-H	$C_{27}H_{21}N_5O$	431	212-214	92
4b	-0	-CH₃	$C_{28}H_{23}N_5O$	445	204-206	89
4c	-0	$-C_2H_5$	$C_{29}H_{25}N_5O$	459	225-228	90
4d	-0	-OCH ₃	$C_{28}H_{23}N_5O_2$	461	247-249	93
4e	-0	-NO ₂	C ₂₇ H ₂₀ N ₆ O3	476	229-231	88
4f	-0	-CI	C ₂₇ H ₂₀ CIN ₅ O	465	236-239	86
4g	-0	-OH	$C_{27}H_{21}N_5O_2$	447	221-223	82
4ĥ	-0	-N(CH ₃) ₂	$C_{29}H_{26}N_6O$	474	224-246	84
4i	-S	-H	$C_{27}H_{21}N_5S$	447	239-242	91
4j	-S	-CH₃	$C_{28}H_{23}N_5S$	461	248-251	86
4k	-S	$-C_2H_5$	$C_{29}H_{25}N_5S$	475	237-239	88
41	-S	-OCH ₃	$C_{28}H_{23}N_5OS$	477	242-244	81
4m	-S	-NO ₂	$C_{27}H_{20}N_6O_2S$	492	245-247	82
4n	-S	-CI	$C_{27}H_{20}CIN_5S$	482	236-238	80
4o	-S	-OH	$C_{27}H_{21}N_5OS$	463	239-242	85
4р	-S	-N(CH3) ₂	$C_{29}H_{26}N_6S$	490	242-245	87

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
4a	-0	-H	N-H peak at 3350cm ⁻¹ ,C=O peak at 1580cm ⁻¹ , Ar C-H peak at 3150cm ⁻¹ , C=N peak at 1670cm ⁻¹ , C-N peak at 1560cm ⁻¹ , N-N peak at 1385cm ⁻¹	432 (M+1)	δ 9.91 (d, $J = 6.8$ Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16 (m, 1H), 7.86 (d, $J = 6.9$ Hz, 1H), 7.67 – 7.61 (m, 2H), 7.50 – 7.34 (m, 5H), 7.31 – 7.25 (m, 1H), 6.30 (dddt, $J = 10.7$, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, $J = 12.3$, 10.2, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, $J = 10.1$, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, $J = 5.3$, 1.7 Hz, 1H), 5.33 (td, $J = 5.7$, 1.7 Hz, 1H), 4.59 – 4.53 (m, 1H), 4.24 (tddd, $J = 7.0$, 4.4, 1.8, 0.9 Hz, 1H), 3.62 (dd, $J = 13.6$, 5.7 Hz, 1H), 3.37 (dd, $J = 13.7$, 5.7 Hz, 1H), 3.10 – 3.02 (m, 1H)
4b	-0	-CH3	N-H peak at 3400cm ⁻¹ ,C=O peak at 1560cm ⁻¹ , Ar C-H peak at 3250cm ⁻¹ , methyl C-H peak at 2800 cm ⁻¹ , C=N peak at 1630cm ⁻¹ , C-N peak at 1550cm ⁻¹ , N-N peak at 1371cm ⁻¹	446 (M+1)	δ 9.91 (d, $J = 6.8$ Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16 (m, 1H), 7.86 (d, $J = 6.9$ Hz, 1H), 7.58 – 7.52 (m, 2H), 7.50 – 7.44 (m, 1H), 7.37 (td, $J = 7.4$, 1.1 Hz, 1H), 7.28 (ddd, $J = 8.2$, 7.2, 1.2 Hz, 1H), 7.19 (dq, $J = 7.9$, 0.8 Hz, 2H), 6.30 (dddt, $J = 10.7$, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, $J = 12.3$, 10.2, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, $J = 10.1$, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, $J = 5.3$, 1.7 Hz, 1H), 5.33 (td, $J = 5.7$, 1.7 Hz, 1H), 4.59 – 4.53 (m, 1H), 4.28 – 4.20 (m, 1H), 3.62 (dd, $J = 13.6$, 5.7 Hz, 1H), 3.37 (dd, $J = 13.7$, 5.7 Hz, 1H), 3.06 (dddt, $J = 8.3$, 5.5, 2.9, 1.7 Hz, 1H)
4c	-0	-C ₂ H ₅	N-H peak at 3406cm ⁻¹ ,C=O peak at 1542cm ⁻¹ , Ar C-H peak at 3031cm ⁻¹ , ethyl C-H peak at 2950 cm ⁻¹ , C=N peak at 1700cm ⁻¹ , C-N peak at 1580cm ⁻¹ , N-N peak at 1368cm ⁻¹	460 (M+1)	$\overline{\delta}$ 9.91 (d, <i>J</i> = 6.8 Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16 (m, 1H), 7.86 (d, <i>J</i> = 6.9 Hz, 1H), 7.58 – 7.52 (m, 2H), 7.50 – 7.44 (m, 1H), 7.37 (td, <i>J</i> = 7.4, 1.1 Hz, 1H), 7.31 – 7.24 (m, 3H), 6.30 (dddt, <i>J</i> = 10.7, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, <i>J</i> = 12.3, 10.2, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, <i>J</i> = 10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, <i>J</i> = 5.3, 1.7 Hz, 1H), 5.33 (td, <i>J</i> = 5.7, 1.7 Hz, 1H), 4.59 – 4.53 (m, 1H), 4.24 (ddddd, <i>J</i> = 7.0, 6.0, 4.4, 1.8, 0.9 Hz, 1H), 3.62 (dd, <i>J</i> =

Table 3. Spectral data of R-substitutedN-((1H-indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carboxamideandN-((1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide indol-3-yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
			·	. , .	13.6, 5.7 Hz, 1H), 3.37 (dd, J = 13.7, 5.7 Hz, 1H),
					3.10 – 3.02 (m, 1H), 2.69 (qt, <i>J</i> = 7.3, 0.9 Hz, 2H),
					1.22 (t, <i>J</i> = 7.2 Hz, 3H)
4d	-0	-OCH ₃	N-H peak at 3300cm ⁻¹ ,C=O peak at 1580cm ⁻¹	462 (M+1)	δ 9.91 (d, <i>J</i> = 6.8 Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16
			', Ar C-H peak at 3250cm ⁻¹ , methyl C-H		(m, 1H), 7.86 (d, <i>J</i> = 6.9 Hz, 1H), 7.65 – 7.59 (m,
			peak at 2850 cm ⁻¹ , C-O peak at 1243 cm ⁻¹ ,		2H), 7.50 – 7.44 (m, 1H), 7.37 (td, <i>J</i> = 7.4, 1.1 Hz,
			C=N peak at 1680cm ⁻¹ , C-N peak at		1H), 7.31 – 7.25 (m, 1H), 6.95 – 6.89 (m, 2H), 6.30
			1596cm ⁻ ', N-N peak at 1385cm ⁻ '		(dddt, J = 10.7, 8.1, 1.9, 0.9 Hz, 1H), 6.12 (ddddd, J)
					= 12.2, 10.3, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, $J =$
					10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, $J = 5.3, 1.7$
					Hz, 1H), 5.33 (td, $J = 5.7, 1.7$ Hz, 1H), 4.59 – 4.53
					(M, 1H), 4.24 (ddddd, J = 7.0, 6.0, 4.4, 1.8, 0.9 HZ, 4.1), 2.00 (a. 201), 2.02 (dd 1, 4.4), 6.0, 5.7 Hz, 4.1)
					$(\Pi), 3.80 (S, 2\Pi), 3.02 (00, J = 13.0, 5.7 \Pi Z, 1\Pi),$
					$3.37 (uu, J = 13.7, 5.7 \Pi Z, 1\Pi), 3.00 (uuul, J = 7.3, 4.7, 1.9, 1.4 \Pi Z, 1\Pi)$
10	0	NO	N H pook at 2250 cm ⁻¹ C-O pook at 1520 cm ⁻¹	A77 (NA+4)	$\frac{4.7, 1.0, 1.1 \Pi 2, \Pi \Pi}{50.01 (d_1 - 6.9 \Pi - 1.1) 0.41 (c_1 \Pi) 0.41}$
46	-0	-INO ₂	1 Ar C-H peak at 3031cm ⁻¹ C-N peak at	477 (101+1)	(m, 3H) 7 90 – 7 83 (m, 3H) 7 50 – 7 44 (m, 1H)
			1730 cm^{-1} C-N peak at 1570 cm^{-1} N-O peak		(11, 51), 7.50 = 7.05 (11, 51), 7.50 = 7.44 (11, 11), 7.37 (td $I = 7.4$ 1 1 Hz 1H) 7.31 = 7.25 (m 1H)
			at 1550 cm ^{-1} N-N peak at 1379cm ^{-1}		6.30 (dddt J = 10.6 8.1 1.8 0.9 Hz 1H) 6.12
					(ddddd, J = 12.3, 10.2, 8.2, 1.7, 0.8 Hz, 2H), 5.96
					(dddd, J = 10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, J =
					5.3, 1.7 Hz, 1H), 5.33 (td, J = 5.7, 1.7 Hz, 1H), 4.59
					- 4.53 (m, 1H), 4.24 (ddddd, <i>J</i> = 7.9, 6.0, 4.4, 1.8,
					0.9 Hz, 1H), 3.62 (dd, <i>J</i> = 13.6, 5.7 Hz, 1H), 3.37
					(dd, <i>J</i> = 13.7, 5.7 Hz, 1H), 3.10 – 3.02 (m, 1H)
4f	-0	-Cl	N-H peak at 3400cm ⁻¹ ,C=O peak at 1570cm ⁻¹	466 (M+1)	δ 9.91 (d, <i>J</i> = 6.8 Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16
			', Ar C-H peak at 3100cm ⁻¹ , C=N peak at		(m, 1H), 7.86 (d, <i>J</i> = 6.9 Hz, 1H), 7.63 – 7.57 (m,
			1700cm ⁻¹ , C-N peak at 1596cm ⁻¹ , C-Cl peak		2H), 7.50 – 7.44 (m, 1H), 7.44 – 7.38 (m, 2H), 7.37
			at 850 cm ⁻⁺ ,N-N peak at 1385cm ⁻⁺		(td, <i>J</i> = 7.4, 1.1 Hz, 1H), 7.31 – 7.25 (m, 1H), 6.30
					(addt, J = 10.7, 8.1, 1.9, 0.9 Hz, 1H), 6.12 (ddddd, J)
					= 12.3, 10.2, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, $J =$
					10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, $J = 5.3, 1.7$
					HZ, 1H), 5.33 (td, $J = 5.7$, 1.7 HZ, 1H), 4.59 – 4.53
					(m, 1H), 4.28 - 4.20 (m, 1H), 3.62 (dd, J = 13.6, 5.7)

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
			·	. , .	Hz, 1H), 3.37 (dd, J = 13.7, 5.7 Hz, 1H), 3.10 – 3.02
					(m, 1H)
4g	-0	-OH	N-H peak at 3450cm ⁻¹ , OH peak at 3200 cm ⁻¹	448 (M+1)	δ 9.91 (d, <i>J</i> = 6.8 Hz, 1H), 9.41 (s, 1H), 8.21 – 8.16
			¹ , C=O peak at 1600cm ⁻¹ , Ar C-H peak at		(m, 1H), 7.86 (d, <i>J</i> = 6.9 Hz, 1H), 7.61 – 7.55 (m,
			3050cm ⁻¹ , C=N peak at 1750cm ⁻¹ , C-N peak		2H), 7.50 – 7.44 (m, 1H), 7.37 (td, <i>J</i> = 7.4, 1.1 Hz,
			at 1520cm ⁻ , N-N peak at 1368cm ⁻		1H), 7.28 (ddd, <i>J</i> = 8.2, 7.1, 1.2 Hz, 1H), 6.88 –
					6.82 (m, 2H), 6.30 (dddt, J = 10.7, 8.1, 1.8, 0.9 Hz,
					1H), 6.12 (ddddd, $J = 12.2, 10.3, 8.2, 1.7, 0.8$ Hz,
					2H), 5.96 (dddd, $J = 10.0, 6.0, 1.8, 0.8$ Hz, 1H),
					5.76 (dq, $J = 5.3$, 1.7 Hz, 1H), 5.33 (td, $J = 5.7$, 1.7
					HZ, 1H), $4.59 - 4.53$ (M, 1H), 4.24 (ddddd, $J = 7.0$,
					$0.0, 4.4, 1.0, 0.9 \ \Pi Z, 1\Pi, 3.0Z (uu, J = 13.0, 5.7)$
					(12, 11), 5.57 (10, 3 = 15.7, 5.7 12, 11), 5.10 - 5.02
<u>4</u> b			N-H peak at 3300 cm^{-1} C=O peak at 1580 cm^{-1}	475 (M+1)	$\frac{111}{5001}$ (d. $l = 6.8 Hz$ 1H) 0.41 (c. 1H) 8.21 8.16
411	-0		1 Ar C-H peak at 3100cm ⁻¹ methyl C-H	475 (101+1)	$(m \ 1H) \ 7\ 86 \ (d \ I - 6\ 9\ Hz \ 1H) \ 7\ 65 - 7\ 59 \ (m$
			p_{car} peak at 2850 cm ⁻¹ C=N peak at 1700cm ⁻¹		(iii, iii), $7.50 - 7.44$ (m 1H) 7.37 (td $J = 7.4 + 1.1$ Hz
			C-N peak at 1596 cm ⁻¹ N-N peak at 1380 cm ⁻¹		1H) $7.31 - 7.25$ (m 1H) $6.77 - 6.72$ (m 2H) 6.30
			1		(dddt, J = 10.7, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, J)
					= 12.2, 10.3, 8.2, 1.7, 0.8 Hz, 2H), 5.96 (dddd, J =
					10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, <i>J</i> = 5.3, 1.7
					Hz, 1H), 5.33 (td, J = 5.7, 1.7 Hz, 1H), 4.59 – 4.53
					(m, 1H), 4.24 (ddddd, <i>J</i> = 7.0, 6.0, 4.4, 1.8, 0.9 Hz,
					1H), 3.62 (dd, <i>J</i> = 13.6, 5.7 Hz, 1H), 3.37 (dd, <i>J</i> =
					13.7, 5.7 Hz, 1H), 3.06 (dddt, <i>J</i> = 7.3, 4.6, 1.8, 1.1
			1		Hz, 1H), 2.92 (s, 5H)
4i	-S	-H	N-H peak at 3450cm ⁻ ,N-C=S peak at	448 (M+1)	δ 9.97 (d, $J = 6.9$ Hz, 1H), 9.83 (s, 1H), 8.21 – 8.16
			2050cm ', Ar C-H peak at 3150cm ', C=N		(m, 1H), 7.85 (d, $J = 6.7$ Hz, 1H), 7.67 – 7.61 (m,
			peak at $1/80$ cm $^{-1}$. C-N peak at 1560 cm $^{-1}$. N-		2H), 7.50 – 7.34 (m, 5H), 7.31 – 7.25 (m, 1H), 6.30
			N peak at 13/1cm		(dddt, J = 10.6, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, J)
					= 12.3, 10.3, 8.4, 1.8, 0.8 HZ, 2H), 5.96 (addd, J = 10.1, 6.0, 1.9, 0.9 Hz, 1H) = 26 (d = 1, 5.2, 4.7)
					$10.1, 0.0, 1.0, 0.0 \Box 2, 1\Box$), $5.70 (00, J = 5.3, 1.7)$
					$(m, 1H), 0.23$ (IU, $J = 0.0, 1.0$ ΠZ , $(\Pi J), 4.03 - 4.30$ (m, 1H) 4.20, 4.20 (m, 1H) 3.60 (dd, $J = 12.6$, 6.0)
					(11, 11), 4.29 - 4.20 (11, 11), 3.00 (uu, J = 13.0, 0.0)

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
					Hz, 1H), 3.35 (dd, J = 13.6, 6.0 Hz, 1H), 3.06 (dddt,
					<i>J</i> = 7.3, 5.3, 2.6, 0.9 Hz, 1H)
4j	-S	-CH₃	N-H peak at 3350cm ⁻¹ ,N-C=S peak at	462 (M+1)	δ 9.97 (d, <i>J</i> = 6.9 Hz, 1H), 9.83 (s, 1H), 8.21 – 8.16
			2100cm ^{-/} , Ar C-H peak at 3000cm ^{-/} , C=N		(m, 1H), 7.85 (d, <i>J</i> = 6.7 Hz, 1H), 7.58 – 7.52 (m,
			peak at 1700cm ', methyl C-H peak at 2800		2H), 7.50 – 7.44 (m, 1H), 7.37 (td, <i>J</i> = 7.4, 1.1 Hz,
			cm ⁻ , C-N peak at 1596cm ⁻ , N-N peak at		1H), 7.31 – 7.25 (m, 1H), 7.22 – 7.16 (m, 2H), 6.30
			1385CM		(dddt, J = 10.6, 8.1, 1.8, 0.9 HZ, 1H), 6.12 (ddddd, J)
					= 12.3, 10.3, 8.4, 1.8, 0.8 HZ, 2H), 5.96 (addd, J = 10.0, 0.0, 1.8, 0.0, 1.5, 1.1), 5.76 (dz, J, 5.2, 1.7)
					$10.0, 0.0, 1.8, 0.8 \Pi Z, 1\Pi$, 5.76 (dq, $J = 5.3, 1.7$
					HZ, HI , 5.23 (I0, $J = 0.0$, 1.8 HZ , HI), 4.03 – 4.08 (m 1H) 4.00 (m 1H) 2.60 (dd J 12.6 6.0
					$(\Pi, \Pi), 4.29 - 4.20 (\Pi, \Pi), 3.00 (uu, J = 13.0, 0.0)$
					(ddda I = 80 AA 18 09 Hz 1H)
14	-9	-C.H	N-H peak at 3400cm ⁻¹ N-C-S peak at	476 (M±1)	$\delta = 0.0, 4.4, 1.0, 0.912, 11)$
40	-0	-02115	2130 cm^{-1} Ar C-H peak at 3150 cm ⁻¹ C=N	470 (MFT)	$(m \ 1H) \ 7 \ 85 \ (d \ J = 6.7 \ Hz \ 1H) \ 7 \ 58 \ - 7 \ 52 \ (m \ 1H) \ 7 \ 58 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ 52 \ - 7 \ - 7 \ 52 \ - 7 \ - 7 \ 52 \ - 7 $
			peak at 1750cm ⁻¹ ethyl C-H peak at 2850		(iii, iii), $7.60 (a, 0 = 0.7 Hz), 110, 7.60 (a, 0 = 0.7 Hz), 110, 7.60 (a, 0 = 0.7 Hz), 110, 7.60 (a, 0 = 0.7 Hz), 110, 100 (a, 0 = 0.7 Hz), 110, 100 (a, 0 = 0.7 Hz), 110, 110, 110, 110, 110, 110, 110, 11$
			cm^{-1} , C-N peak at 1585 cm^{-1} , N-N peak at		1H), $7.31 - 7.24$ (m, 3H), 6.30 (dddt, $J = 10.6, 8.1$,
			1368cm ⁻¹		1.8. 0.9 Hz. 1H), 6.12 (ddddd, $J = 12.3, 10.3, 8.4$.
					1.8, 0.8 Hz, 2H), 5.96 (dddd, J = 10.0, 6.0, 1.8, 0.8
					Hz, 1H), 5.76 (dq, J = 5.3, 1.7 Hz, 1H), 5.23 (td, J =
					6.0, 1.8 Hz, 1H), 4.63 – 4.58 (m, 1H), 4.29 – 4.20
					(m, 1H), 3.60 (dd, <i>J</i> = 13.6, 6.0 Hz, 1H), 3.35 (dd, <i>J</i>
					= 13.6, 6.0 Hz, 1H), 3.06 (dddq, <i>J</i> = 8.0, 4.4, 1.9,
					0.9 Hz, 1H), 2.69 (qt, <i>J</i> = 7.3, 0.9 Hz, 2H), 1.22 (t, <i>J</i>
			1		= 7.2 Hz, 3H)
41	-S	-OCH ₃	N-H peak at 3350cm ⁻¹ ,N-C=S peak at	478 (M+1)	δ 9.97 (d, $J = 7.0$ Hz, 1H), 9.83 (s, 1H), 8.21 – 8.16
			2100cm ⁻ , Ar C-H peak at 3150cm ⁻ , C-O		(m, 1H), 7.85 (d, $J = 6.8$ Hz, 1H), 7.65 – 7.59 (m,
			peak at 1240 cm ⁻ , C=N peak at 1698cm ⁻ ,		2H), $7.50 - 7.44$ (m, 1H), 7.37 (td, $J = 7.4$, 1.1 Hz,
			methyl C-H peak at 2800 cm $^{-1}$, C-N peak at		(11), 7.31 - 7.25 (m, 1H), 6.95 - 6.89 (m, 2H), 6.30 (dd dd 1, 200 - 2
			1596cm, N-N peak at 1379cm		$(0001, J = 10.0, 0.1, 1.0, 0.9 \Pi Z, 1\Pi), 0.12 (00000, J = 12.4, 10.2, 9.4, 1.9, 0.9 \Pi Z, 2\Pi), 5.06 (dddd, J = 1.2, 1.0, 0.1, 1.0, 0.9, 0.9, 0.9, 0.9, 0.9, 0.9, 0.9, 0$
					$= 12.4, 10.3, 0.4, 1.0, 0.0 \ \Pi Z, 2\Pi J, 3.90 (0000, J = 10.1, 6.0, 1.8, 0.8 \ Hz, 1H) 5.76 (do, 1 - 5.2, 1.7)$
					Hz 1H) 5 23 (td $J = 6.0, 1.8$ Hz 1H) $4.63 = 4.58$
					$(m \ 1H) \ 4 \ 29 = 4 \ 20 \ (m \ 1H) \ 3 \ 80 \ (s \ 2H) \ 3 \ 60 \ (dd)$
					(11, 11), 4.20 - 4.20 (11, 11), 5.00 (3, 21), 5.00 (dd, 10)

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
					J = 13.5, 6.0 Hz, 1H), 3.35 (dd, J = 13.7, 6.0 Hz,
					1H), 3.06 (ddddd, <i>J</i> = 8.4, 4.6, 2.8, 1.7, 1.0 Hz, 1H)
4m	-S	-NO ₂	N-H peak at 3350cm ⁻¹ ,N-C=S peak at	493 (M+1)	δ 9.97 (d, <i>J</i> = 6.9 Hz, 1H), 9.83 (s, 1H), 8.21 – 8.15
			2150cm ⁻¹ , Ar C-H peak at 3100cm ⁻¹ , C=N		(m, 3H), 7.90 – 7.82 (m, 3H), 7.50 – 7.44 (m, 1H),
			peak at 1720cm ⁻¹ , C-N peak at 1570cm ⁻¹ , N-		7.37 (td, <i>J</i> = 7.4, 1.1 Hz, 1H), 7.31 – 7.25 (m, 1H),
			O peak at 1550 cm ⁻⁺ , N-N peak at 1371cm ⁻⁺		6.30 (dddt, <i>J</i> = 10.6, 8.1, 1.9, 0.9 Hz, 1H), 6.12
					(ddddd, <i>J</i> = 12.3, 10.3, 8.4, 1.8, 0.8 Hz, 2H), 5.96
					(dddd, <i>J</i> = 10.0, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, <i>J</i> =
					5.3, 1.7 Hz, 1H), 5.23 (td, $J = 6.0$, 1.8 Hz, 1H), 4.63
					-4.58 (m, 1H), $4.29 - 4.20$ (m, 1H), 3.60 (dd, $J =$
					13.6, 6.0 Hz, 1H), 3.35 (dd, $J = 13.6$, 6.0 Hz, 1H),
					3.06 (dddt, J = 7.3, 5.3, 2.6, 0.9 HZ, 1H)
4n	-8	-CI	N-H peak at 3400cm ⁻ ,N-C=S peak at	483 (M+1)	0.9.97 (d, $J = 6.9$ Hz, 1H), 9.83 (s, 1H), $8.21 - 8.16$
			2050 cm , Ar C-H peak at 3050 cm , C=N		(M, 1H), 7.85 (d, J = 6.7 HZ, 1H), 7.63 - 7.57 (M, 0H) = 7.44 (m
			peak at 1780 cm ⁻¹ N N peak at 1585 cm ⁻¹		$Z\Pi$), 7.50 – 7.44 (III, 1H), 7.44 – 7.38 (III, 2H), 7.37 (td. 1, 7.4, 4.4 Hz, 4H), 7.24, 7.25 (m. 4H), 6.20
			Cipeak at 800 cm , N-N peak at 1385cm		$(IU, J = 7.4, I.1 \square Z, I\square), 7.31 - 7.23 (III, I\square), 0.30$
					(0001, 5 = 10.0, 0.1, 1.9, 0.9112, 111), 0.12 (00000, 5 = 12.2, 10.2, 9.4, 1.9, 0.9 Hz, 2H) = 5.06 (dddd, 1 = 12.2, 10.2, 9.4, 1.9, 0.9 Hz, 2H) = 5.06 (dddd, 1 = 12.2, 10.2, 9.4, 1.9, 0.9 Hz, 2H) = 5.06 (dddd, 1 = 12.2, 10.2, 9.4, 1.9, 0.9 Hz, 2H)
					= 12.3, 10.3, 0.4, 1.0, 0.012, 211, 3.50 (dddd, 0 = 10.0, 60, 1.8, 0.8 Hz, 1H), 5.76 (dg, 1 = 5.3, 1.7)
					H_{7} (14), 5.23 (td. $J = 6.0, 1.8$ Hz, 1H), 4.63 – 4.58
					$(m \ 1H) \ 4 \ 25 \ (dtdd \ J = 7 \ 9 \ 5 \ 9 \ 17 \ 0 \ 9 \ Hz \ 1H)$
					3.60 (dd, J = 13.6, 6.0 Hz, 1H), 3.35 (dd, J = 13.6, 5.0 Hz, 5.0
					6.0 Hz, 1H), 3.06 (dddt, $J = 7.3, 5.4, 2.6, 0.9 Hz$.
					1H)
40	-S	-OH	N-H peak at 3250cm ⁻¹ ,N-C=S peak at	464 (M+1)	δ 9.97 (d, J = 6.9 Hz, 1H), 9.83 (s, 1H), 8.21 – 8.16
			2150cm ⁻¹ , OH peak at 3400 cm ⁻¹ , Ar C-H		(m, 1H), 7.85 (d, J = 6.7 Hz, 1H), 7.61 – 7.55 (m,
			peak at 3150cm ⁻¹ , C=N peak at 1700cm ⁻¹ , C-		2H), 7.50 – 7.44 (m, 1H), 7.37 (td, <i>J</i> = 7.4, 1.1 Hz,
			N peak at 1590 cm ⁻¹ , C-Cl peak at 800 cm ⁻¹ ,		1H), 7.28 (ddd, <i>J</i> = 8.2, 7.1, 1.2 Hz, 1H), 6.88 –
			N-N peak at 1375cm ⁻¹		6.82 (m, 2H), 6.30 (dddt, <i>J</i> = 10.6, 8.1, 1.8, 0.9 Hz,
					1H), 6.12 (ddddd, <i>J</i> = 12.3, 10.3, 8.4, 1.8, 0.8 Hz,
					2H), 5.96 (dddd, <i>J</i> = 10.1, 6.0, 1.8, 0.8 Hz, 1H),
					5.76 (dq, <i>J</i> = 5.3, 1.7 Hz, 1H), 5.23 (td, <i>J</i> = 6.0, 1.8
					Hz, 1H), 4.63 – 4.58 (m, 1H), 4.29 – 4.20 (m, 1H),
					3.60 (dd, <i>J</i> = 13.6, 6.0 Hz, 1H), 3.35 (dd, <i>J</i> = 13.7,

Comp	-X	-R	IR Spectra	Mass spectra(m/z)	¹ H NMR spectra (DMSO)
					6.0 Hz, 1H), 3.06 (dddq, <i>J</i> = 8.2, 4.6, 2.0, 1.0 Hz,
					1H)
4p	-S	-N(CH ₃) ₂	N-H peak at 3300 cm ⁻¹ , N-C=S peak at 2100 cm ⁻¹ . Ar C H peak at 2150 cm ⁻¹ . C-N	491 (M+1)	δ 9.97 (d, J = 7.0 Hz, 1H), 9.83 (s, 1H), 8.21 – 8.16 (m, 1H), 7.85 (d, J = 6.7 Hz, 1H), 7.65 – 7.50 (m, 1H)
			peak at 1750 cm^{-1} , methyl C-H peak at 2800		(iii, 11), 7.65 (d, $J = 0.712$, 11), 7.65 – 7.59 (iii, 2H), 7.50 – 7.44 (m, 1H), 7.37 (td, $J = 7.4$, 1.1 Hz,
			cm ⁻¹ , C-N peak at 1600cm ⁻¹ , N-N peak at		1H), 7.31 – 7.25 (m, 1H), 6.78 – 6.72 (m, 2H), 6.30
			1379cm ⁻¹		(dddt, J = 10.6, 8.1, 1.8, 0.9 Hz, 1H), 6.12 (ddddd, J
					= 12.4, 10.4, 8.4, 1.8, 0.8 Hz, 2H), 5.96 (dddd, <i>J</i> =
					10.1, 6.0, 1.8, 0.8 Hz, 1H), 5.76 (dq, <i>J</i> = 5.3, 1.7
					Hz, 1H), 5.23 (td, <i>J</i> = 6.0, 1.8 Hz, 1H), 4.63 – 4.58
					(m, 1H), 4.29 – 4.20 (m, 1H), 3.60 (dd, <i>J</i> = 13.5, 6.0
					Hz, 1H), 3.35 (dd, <i>J</i> = 13.6, 6.0 Hz, 1H), 3.06
					(ddddd, <i>J</i> = 8.3, 4.6, 2.8, 1.7, 1.0 Hz, 1H), 2.92 (s,
					5H)

3.2 Anti-Diabetic Activity

3.2.1 In vivo anti-diabetic activity

Effect of Synthesized Compounds on Body Weight: Table 4 presents the effect of synthesized compounds and glibenclamide on changes in body weight. In diabetic control rats, there was a significant decrease (22.57%) in final body weight when compared to normal control rats. Treatment with synthesized compounds showed significant (P < 0.05) increases in body weight of diabetic rats. Change in body weight was minimal for positive control (1.02%) whereas synthesized compounds treatments showed moderate improvement in body weight (1.50– 2.51%).

Effect of Synthesized Compounds on Glycosylated Hemoglobin (HbA1c): Fig. 1 shows the effect of synthesized compounds and

glibenclamide on HbA1c level in normal and experimental rats. STZ treated rats showed a significant elevation in HbA1c level (5.25%) as compared with normal control. Following synthesized compounds and glibenclamide administration to diabetic rats caused a significant reduction (P < 0.05) in HbA1c level (~3–5%) as compared to diabetic control rats.

Effect of Synthesized Compounds on Blood Glucose: The blood glucose level was measured in normal and experimental groups at 0 days, 7th day, 14th and 21st day of treatment. STZ administration showed a significant increase (P < 0.05) in the blood glucose level when compared to normal control group. There was dramatically reduction in blood glucose level from 221 mg/dL to 188 mg/dL after 21 days of treatment with synthesized compound 4a at the dose of 50 mg/kg b.wt (Table 5).

 Table 4. Effect of synthesized compounds and glibenclamide on body weight and glycosylated hemoglobin content in STZ-induced diabetic rats

Groups	Body V	Veight (g)	% Change in	HbA1c (%)
	Initial	Final	body weight	
Group-I, Normal Control	198 ± 4.20	204 ± 3.42	2.51	4.90 ± 0.22
Group-II, Diabetic control	196 ± 5.65	151 ± 4.59a	-22.57	10.23 ± 0.27
Group-III, Diabetic + synthesized compound 4a ,(50 mg/kg b.wt)	192 ± 4.80	195 ± 3.62b	1.52	6.34 ± 0.20
Group-IV,Diabetic + synthesized compound 4m,(50 mg/kg b.wt)	194 ± 2.30	198 ± 3.30b	2.51	7.92 ± 0.18
Group-V,Diabetic+Glibenclamide (5 mg/kg b.wt)	195 ± 4.90	197 ± 4.35b	1.025	6.49 ± 0.16

Data represented as mean ± SEM (n=6)

a = (P < 0.05) statistically significant difference when compared with Normal control

b = (P < 0.05) statistically significant difference when compared with Diabetic control





Table 5.	Effect of synthesized compounds and glibenclamide on blood glucose level in STZ-				
induced diabetic rats					
•	Discut where a state of the set o				

Groups	Blood glucose level (mg/dL)					
	•	Time inte	ervals (Da	ays)		
	0	7	14	21		
Group-I Normal Control	80	79	81	78		
Group-II, Diabetic control	276	310	320	315		
Group-III, Diabetic + synthesized compound 4a ,(50 mg/kg	272	250	221	188		
D.wt)						
Group-IV, Diabetic + synthesized compound 4m ,(50 mg/kg	274	258	236	211		
b.wt)						
Group-V, Diabetic+Glibenclamide ,(5 mg/kg b.wt)	276	248	215	175		





3.2.2 In vitro anti-diabetic activity

In vitro α -amylase Inhibition Assay: In the present study, synthesized compounds showed a significant inhibition of α - amylase enzyme activity in a concentration dependent manner. Synthesized compounds 4a, 4e and 4m at the concentrations 20, 40, 60, 80 and 100 µg/ml gives different % inhibition of α -amylase enzyme activity. The acarbose used as a reference standard at the same concentrations showed variable inhibition of α -amylase activity (Table 6).

In vitro α-Glucosidase Inhibition Assay: In the present study, synthesized compounds showed a significant inhibition of α-Glucosidase enzyme activity in a concentration dependent manner. Synthesized compounds 4a, 4e and 4m at the concentrations 20, 40, 60, 80 and 100 µg/ml ml gives different % inhibition α-Glucosidase enzyme activity. The acarbose used as a reference standard at the same concentrations showed variable inhibition of α-Glucosidase activity (Table 7).

Table 6. Effect of synthesized compounds and acarbose in the in vitro α -amylase inhibition
model

S. No	Synthesized compound	% inhibition of α-amylase enzyme activity concentration of sample (μg/ml)					
		20	40	60	80	100	
1	4a	24	46	58	66	70	
2	4e	22	42	50	58	68	
3	4m	18	39	46	55	60	
4	Acarbose	27	51	64	82	87	



Fig. 3. Effect of Synthesized compounds 4a, 4e, 4m and acarbose in the *in vitro* α -amylase inhibition model

Table 7. Effect of synthesized compounds and acarbose in the *in vitro* α-Glucosidase inhibition model

S. No	Synthesized compound	% inhibition of α-Glucosidase enzyme activity concentration of sample (μg/ml)					
		20	40	60	80	100	
1	4a	12	24	36	55	66	
2	4e	9	22	33	41	59	
3	4m	8	17	26	34	46	
4	Acarbose	15	32	48	64	75	





4. CONCLUSION

The present study was aimed to synthesis, characterization of the pyrazoline fused indole derivatives. The synthesis of novel compounds involves the four steps; in the first step indole was treated with N, N-dimethyl formamide to synthesize the compound 1 indole-3-aldehyde. In the second step compound 1 was treated with 4-Substituted acetophenone to synthesize compound 2(a-h). In the third step chalcones (2a-2h) were treated with thiosemicarbazide or

semicarbazide to synthesize compound 3(a-p). In the final step compound 3(a-p) were treated with indole-3-aldehyde to prepare the final product RsubstitutedN-((1H-indol-3-yl) methylene)-5-(1Hindol-3-yl)-3-phenyl-4,5-dihydro-1H-pyrazole-1carboxamide and R-substitutedN-((1H-indol-3yl)methylene)-5-(1H-indol-3-yl)-3-phenyl-4,5dihydro-1H-pyrazole-1-carbothioamide 4(a-p). The chemical structures of the synthesized compounds were characterized by means of IR, Mass and NMR spectroscopy. The compounds were screened for anti-diabetic activity by In-vitro and In-vivo methods. In In-vivo method 4a, 4m have exhibited moderate anti-diabetic activity as that of standard drug, glibenclamide. In In-vitro method 4a, 4e & 4m have shows moderate antidiabetic activity as that of reference drug, acarbose. These compounds can be further exploited to get the potent lead compound.

CONSENT

It is not applicable.

ETHICAL APPROVAL

Animal Ethic committee approval has been taken to carry out this study.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

REFERENCES

- Kumar S, Bawa S, Drabu S, Kumar R, Gupta H. Biological activities of pyrazoline derivatives--a recent development. Recent Pat Antiinfect Drug Discov [jour]. 2009;4(3):154-63. DOI: 10.2174/157489109789318569
- 2. Allan and Richmond synthesis of novel heterocyclic substances Journal of Organic Chemistry 1998;2(22).
- 3. Dias LRS, salvador RRS, pyrazole carbahydrazole derivatives of pharmaceutical interest. Pharmaceuticals. 2012;5:317-24.
- Lakhdar S, Westermaier M, Terrier F, Goumont R, Boubaker T, Ofial AR, Mayr H. Nucleophilic reactivities of indoles. J Org Chem. 2006;71:9088–9095.
- 5. Sharma V, Pradeep K, Devender P. Biological importance of the indole nucleus in recent years: a comprehensive review. J Heterocycl Chem. 2010;47:491–502.
- Kaushik NK, Kaushik N, Attri P, Kumar N, Kim CH, Verma AK, Choi EH. Biomedical Importance of Indoles. Molecules. 2013;18:6620–6662.
- Xue S, Ma L, Gao R, Lin Y, Linn Z. Synthesis and antiviral activity of some novel indole-2-carboxylate derivatives. Acta Pharmaceutica Sinica B. 2014;4(4):313–321.
- 8. Pedada SR, Yarla NS, Tambade PJ, Dhananjaya BL, Bishayee A, Arunasree KM, et al. Synthesis of new secretory

phospholipase A2-inhibitory indole containing isoxazole derivatives as antiinflammatory and anticancer agents. Eur J Med Chem. 2016;112:289–297.

- 9. Zhuang SH, Lin YC, Chou LC, Hsu MH, Lin HY, Huang CH, et al. Synthesis and anticancer activi ty of 2, 4-disubstituted furo[3,2-b]indole derivatives. Eur J Med Chem. 2013;66:466–479.
- Kasralikar HM, Jadhavar SC, Bhusare SR. Synthesis and molecular docking studies of oxochromenyl xanthenone and indolyl xanthenone derivatives as anti-HIV-1 RT inhibitors. Bioorg Med Chem Lett. 2015;25:3882–3886.
- Silveira CC, Mendes SR, Soares JR, Victoria FN, Martinez DM, Savegnago L. Synthesis and antioxidant activity of new C-3 sulfenyl indoles. Tetrahedron Lett. 2013;54:4926–4929.
- Xu H, Fan LL. Antifungal agents. Part 4: Synthesis and antifungal activities of novel indole[1,2-c]-1,2,4-benzotriazine derivatives against phytopathogenic fungi invitro. Eur J Med Chem. 2011;46:364– 369.
- Velezheva V, Brennan P, Ivanov P, Kornienko A, Lyubimov S, Kazarian K, et al. Synthesis and antituberculosis activity of indole–pyridine derived hydrazides, hydrazide–hydrazones, and thiosemicarbazones. Bioorg Med Chem Lett. 2016;26(3):978–985.
- Tejasree, et al. Antidiabetic Activity Of 1-(4-(Dimethylamino)Benzylidene)-5-(2-Oxo Indolin-3-Ylidene) Thiocarbohydrazone In Rats. IJPSR. 2014;5(7):2738-2743.
- 15. Luthra T, Nayak AK, Bose S, Chakrabarti S, Gupta A, Sen S. Indole based antimalarial compounds targeting the melatonin pathway: their design, synthesis and biological evaluation. Eur J Med Chem. 2019;168:11–27.
- Bingul M, Ercan S, Boga M. The design of novel 4, 6-dimethoxyindole based hydrazide-hydrazones: Molecular modeling, synthesis and anticholinesterase activity. J Mol Struct. 2020;1213:128202.
- 17. The Fischer indole synthesis Philip A. Roussel J. Chem. Educ. 1953;30(3):122.
- 18. Panther J, Müller TJJ. Synthesis. 2016;48:974-986.
- Rhodium(III)-Catalyzed Indole Synthesis Using N–N Bond as an Internal Oxidant by B. Liu, C. Song, C. Sun, S. Zhou, J. Zhu, J. Am. Chem. Soc. 2013;135:16625-16631.

- 20. Total Synthesis of Indoles from Tricholoma Species via Bartoli/Heteroaryl Radical Methodologies by A. Dobbs, J. Org. Chem. 2001;66:638-641.
- Pd-Catalyzed Cascade Reaction for the Synthesis of 2-Substituted Indoles by J. Jadhav, V. Gaikwad, R. Kurane, R. Salunkhe, G. Rahsinkar, Synlett. 2012;23:2511-2515.
- 22. Gosh MN. Toxicity studies, Fundamentals of Experimental Pharmacology, Scientific Book Agency, Calcutta. 1984;153–158.
- 23. Gajdosik A, Gajdosikova M, Stefek J. Navarova R. Hozova, Streptozotocininduced experimental

diabetes in male Wistar rats, Gen. Physiol. Biophys. 1999;18:54–62.

- 24. Nayak SS, Pattabiraman TN. A new colorimetric method for the estimation of glycosylated hemoglobin, Clin. Chim. Acta. 1981;109:267–274.
- 25. Kuppusamy A, Muthusamy U, Andichetiar S. Thirumalaisamy S, Varadharajan K, Ramasamy, et al., *In vitro* (α -glucosidase and α -amylase inhibition) and in vivo antidiabetic property of phytic acid (IP6) in streptozotocin-nicotinamideinduced type 2 diabetes mellitus (NIDDM) in rats, Journal of Complementary and Integrative Medicine. 2011;8(1):9.

© 2021 Vaddiraju et al.; This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Peer-review history: The peer review history for this paper can be accessed here: https://www.sdiarticle5.com/review-history/78061